



# BOOK OF ABSTRACTS

## **The 3<sup>rd</sup> International Conference on Nanomaterials: Fundamentals and Applications**

October 09-11, 2017, Štrbské Pleso, High Tatras, Hotel Patria,  
Slovak republic

Organized by:

Department of Physical Chemistry  
Faculty of Natural Sciences  
Pavol Jozef Safarik University in Kosice  
&  
Slovak Chemical Society  
Bratislava

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**Nanoceramic composites based on organomagnesiumoxanealumoxanes**

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Nanostructured ceramics have unique properties and performance characteristics. However, monolithic nanoceramics are inferior to nanoceramic composite materials, reinforced by continuous or discontinuous ceramic fibers, according to some physical-mechanical parameters, for example, in crack resistance (fracture toughness). The use of discontinuous fibers for reinforcing ceramics has a number of significant advantages, such as precise control of the fiber amount, fabrication cost reduction, and simpler production of irregular shapes. However, a disadvantage of composites reinforced with discontinuous fibers is that they have lower fracture toughness, as compared to composites reinforced with continuous fibers. To improve this parameter, the most uniform mixing of the chopped fibers with the matrix is required.

Another possible method to avoid the problem of homogeneous dispersion of fibers within the matrix is self-reinforcement of the composite [1].

We obtained nanoceramic composites (Fig. 1) from powders of nano-SiC, micro-SiC and a binder based on organomagnesiumoxanealumoxanes [2] (precursors of aluminum-magnesium spinel) first synthesized in the State Research Institute for Chemistry and Technology of Organoelement Compounds, and reinforcing discontinuous spinel fibers of  $\text{MgAl}_2\text{O}_4$  composition. Spinel fibers were obtained by forming polymer fibers from fiber-forming organomagnesiumoxanealumoxanes [2, 3], followed by their pyrolysis (Fig. 2). It is known that due to the high melting point (2135 °C), hardness, chemical resistance and strength, which remain at high temperatures, aluminum-magnesium spinel is an effective material for high-temperature technology.

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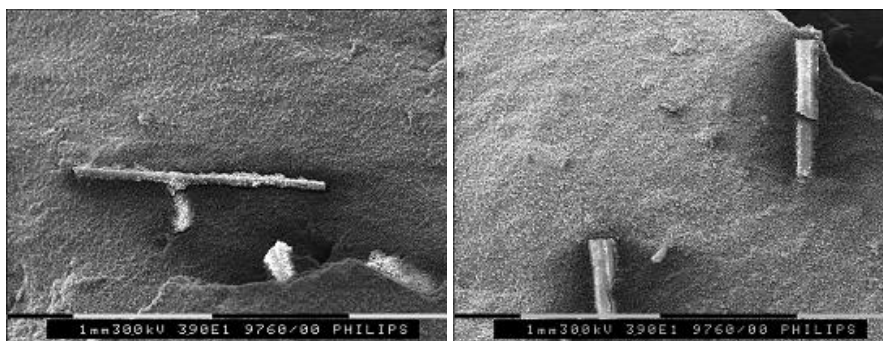


Fig.1 – SEM of the nanoceramic composite

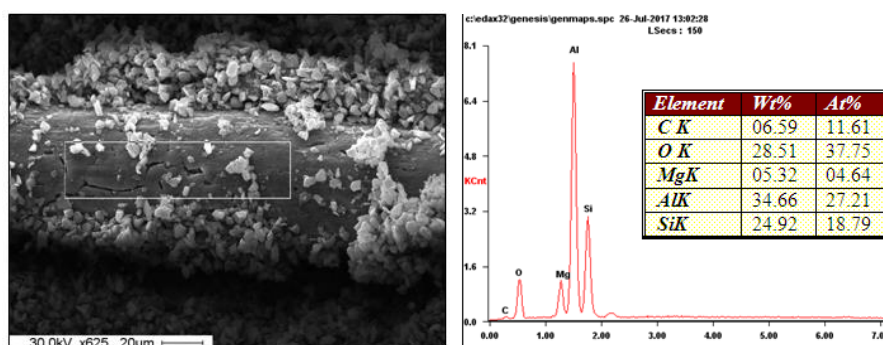


Fig. 2 – SEM and X-ray elemental microanalysis of spinel fiber in a ceramic composite

Moreover, the formation of nanofibers composed of  $\text{Si}_3\text{N}_4$  was observed. That is, self-reinforcement of the nanoceramic composite occurred (Fig. 3), which positively effected its physical and mechanical properties (density, strength, fracture toughness).

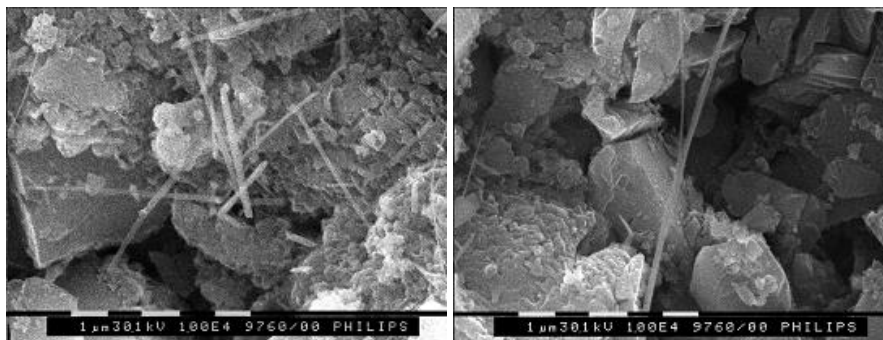


Fig. 2 – SEM of self-reinforced nanoceramic composite

Thus, organomagnesiumoxanealumoxanes can be effectively used as precursors of components (binders, fibers, powders) for high-strength and high-temperature nanoceramic composites.

#### Acknowledgements

This work was supported by the Russian Foundation for Basic Research, project no. 17-03-00331 A.

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**SESSION IV Advanced preparation methods and characterization techniques of nanomaterials**

**Nano-oxide ceramics based on organoelementoxanes**

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The ceramic-forming organoelement poly(oligo)mers are a new class of promising precursors for obtaining high-temperature and oxidation-resistant components (ceramic fibers, matrixes, combined protective and barrier coatings, powders) of nanostructured ceramic composites. They allow to apply radically different “polymer technology” for ceramics production through pyrolysis of the ceramic-forming organoelement poly(oligo)mers [1].

On the basis of unstable chemically active organoaluminum compounds we have developed a method for the synthesis of hydrolytically stable in air ceramic-forming organoalumoxane oligomers - chelated alkoxyalumoxanes [2].

It is found that the thermochemical treatment of ceramic-forming organoalumoxanes results in corundum ceramics  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> [2].

Based on such ceramic-forming organoalumoxanes, a process for the preparation of a high-purity nanostructured, silica free organoalumoxane binder of corundum composition  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> was developed [3].

The ceramic-forming organoalumoxanes were found to readily interact with metal alkoxides and acetylacetonates and they may comprise a wide range of elements and metals in various combinations and ratios with homogeneous distribution in oligomer matrix on a molecular level. This provides a perfect nanostructure, enhanced physical-mechanical and operational properties of ceramic composites obtained on the basis of ceramic-forming organoelementoxanes. The organometallic synthesis ensures high purity of the starting oligomers. This allows obtaining high-purity oxides of elements of a given composition and producing materials on their base at much lower temperatures as compared with technical-grade oxides use [1].

The hydrolytically stable in air, soluble in organic solvents ceramic-forming organoyttriumoxanealumoxanes having fiber-forming properties were synthesized by the co-condensation of chelated organoalumoxane oligomers and yttrium acetylacetonate [4,5].

Thermochemical transformation (pyrolysis) of ceramic-forming organoyttriumoxanealumoxanes at 1500 °C results in aluminum-yttrium ceramics.

It is found that the pyrolysis of organoyttriumoxanealumoxanes with Al:Y $\approx$ 1.5-1.7 leads to the formation of an aluminum-yttrium garnet, the samples are a ceramic matrix consisting of globular formations with nanostructured elements whose characteristic dimensions do not exceed 10 nm.

Polymeric organoaluminum-yttrium fibers based on organoyttriumoxanealumoxanes in the pyrolysis process are transformed into ceramic refractory oxide fibers based on alumina, modified with yttrium oxides, including fibers of aluminum-yttrium garnet Al<sub>5</sub>Y<sub>3</sub>O<sub>12</sub> composition [5].

The hydrolytically stable in air, soluble in organic solvents ceramic-forming organomagnesiumoxanealumoxanes having fiber-forming properties were synthesized by the chelated alkoxyalumoxane interaction with magnesium acetylacetonate [6].

We found that organomagnesiumoxanealumoxanes pyrolysis with molar ratio of Al:Mg = 2:1 even at 900 °C results in magnesium-aluminate MgAl<sub>2</sub>O<sub>4</sub>.

Based on ceramic-forming organoelementoxanes (organoalumoxanes, organoyttriumoxanealumoxanes, organomagnesiumoxanealumoxanes) the researchers of SSC RF GNIChTEOS developed effective method for obtaining components of heat - resistant ceramics and ceramic composites of oxide composition (Al<sub>2</sub>O<sub>3</sub>, Y<sub>2</sub>O<sub>3</sub>–Al<sub>2</sub>O<sub>3</sub>, MgO–Al<sub>2</sub>O<sub>3</sub>): binders, impregnation compositions, ceramic nanopowders [1,3,6,7].

**Acknowledgements**

This work was supported by the Russian Foundation for Basic Research, project no. 17-03-00331 A.

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